

Reaction of Methylene-cyclopropanes with Palladium Chloride

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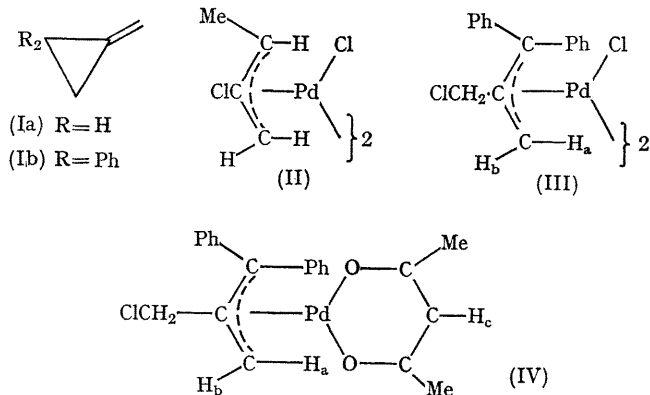
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Summary The preparation of π -allylpalladium complexes from methylenecyclopropanes and palladium chloride is described.

4.00 (d, H_b, J 1.2 Hz), 4.03 and 4.20 (AB qu, J_{AB} 10.8 Hz, CH₂Cl), 5.37 (s, H_c), and 7.1—7.7 p.p.m. (m, C₆H₅).

MUCH attention has been focused on the reaction of cyclopropane derivatives with transition-metal complexes.¹ We have examined the reaction with palladium chloride.

Methylenecyclopropane (Ia) in benzene was added with stirring (room temp.) to dichloro-bis(benzonitrile)palladium in the same solvent. The resulting yellow crystals were recrystallised from benzene-*n*-hexane (1:4) to give di- μ -chloro-bis(2-chloro-1-methyl- π -allyl)dipalladium(II) (II) in nearly quantitative yields, m.p. 179—183° (decomp.) (lit.,² 182—186°). The spectral properties were in agreement with those of an authentic sample. Under the same conditions, 2,2-diphenylmethylenecyclopropane (Ib) afforded di- μ -chloro-bis(2-chloromethyl-1,1-diphenyl- π -allyl)dipalladium(II) (III) [80%, m.p. 208—210° (decomp.)]. The i.r. spectrum (tetrachloroethylene suspension) indicated the absence of methyl and terminal methylene groups. The n.m.r. spectrum (CDCl₃, δ from Me₄Si): 3.52 (br s, H_a, $w_{1/2}$ 3.3 Hz), 4.16 (d, H_b, J 1.5 Hz), 4.07 and 4.31 (AB qu, J_{AB} 11.4 Hz, CH₂Cl), and 7.1—7.7 p.p.m. (m, C₆H₅). The assigned structure confirmed by reduction (H₂, 60°) to give 1,1-dimethyl-2,2-diphenylethylene (65%). On treatment with thallos acetylacetonate, (III) was converted to mononuclear acetylacetonato(2-chloromethyl-1,1-diphenyl- π -allyl)palladium(II) (IV), [85%, m.p. 55—58°; n.m.r. (CDCl₃): δ 1.99 and 2.05 (2 s, CH₃), 3.39 (br s, H_a, 3.3 Hz),



Hence the reaction leads to the formation of π -allylic palladium complexes. The direction of ring fission is markedly influenced by substitution. The reaction of methylenecyclopropane itself (Ia) involves the fission of the C-1-C-2 bond, (*cf.* the reaction of vinylcyclopropane derivatives with palladium chloride³). The diphenyl derivative (Ib), however, gives the C-2-C-3 cleavage product. These findings are reminiscent of the behaviour of methylenecyclopropanes towards di-iron enneacarbonyl.⁴

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¹ E.g., C. F. H. Tipper, *Proc. Chem. Soc.*, 1955, 713; D. M. Adams, J. Chatt, and R. G. Guy, *ibid.*, 1960, 179; D. M. Adams, J. Chatt, R. G. Guy, and N. Sheppard, *J. Chem. Soc.*, 1961, 738; S. Sarel, R. Ben-Shoshan, and B. Kirson, *J. Amer. Chem. Soc.*, 1965, 87, 2517; C. H. DePuy, V. M. Kobel, and D. H. Gibson, *J. Organometallic Chem.*, 1968, 13, 266; A. D. Ketley and J. A. Braatz, *Chem. Comm.*, 1968, 959; W. J. Irwin and F. J. McQuillin, *Tetrahedron Letters*, 1968, 1937.

² M. S. Lupin, J. Powell, and B. L. Shaw, *J. Chem. Soc. (A)*, 1966, 1687.

³ T. Shono, T. Yoshimura, Y. Matsumura, and R. Oda, *J. Org. Chem.*, 1968, 33, 876.

⁴ R. Noyori, T. Nishimura, and H. Takaya, *Chem. Comm.*, 1969, 89.